

Correction

Vapor–Liquid Equilibria for the Binary Systems of Methylcyclohexane with 1-Propanol, 2-Propanol, 1-Butanol, and 2-Butanol at 101.3 kPa. Sonia Loras, Juan B. Montón, and Francisco España, *J. Chem. Eng. Data* 1997, 42, 914–918.

In Table 1 the density of 2-butanol in the literature column is wrong; it should be 802.44 kg·m⁻³. The correct table follows.

Table 1. Physical Properties of Chemicals: Densities d and Refractive Indexes n at 298.15 K and Normal Boiling Points T_b of Pure Compounds

compound	$d/(\text{kg}\cdot\text{m}^{-3})$		n		T_b/K	
	exptl	lit.	exptl	lit.	exptl	lit.
1-butanol	805.64	805.60 ^a	1.3972	1.3972 ^a	390.5	390.55 ^a
2-butanol	802.62	802.44 ^a	1.3950	1.3950 ^a	372.1	372.35 ^a
1-propanol	799.51	799.50 ^a	1.3830	1.3831 ^a	369.9	370.01 ^a
2-propanol	780.95	780.93 ^a	1.3748	1.3750 ^a	355.1	355.39 ^a
methylcyclohexane	764.83	764.59 ^a	1.4194	1.4206 ^b	373.6	374.08 ^a

^a TRC Thermodynamic Tables Hydrocarbons, 1996. ^b Data Compilation Tables of Properties of Pure Compounds (Daubert and Danner, 1989).

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